## Acta Cryst. (1957). 10, 708

The double circular-disc sector used for electron diffraction. By Mieko Takagi, Tokyo Institute of<br>Technology, Oh-Okayama, Tokyo, Japan

(Received 28 June 1957)

Since the electron-diffraction method is increasing its importance in the structure determination of crystals and of amorphous substances, as well as of gas molecules, there is a growing demand for accuracy of intensity measurement of scattered electrons. In the study of gas molecules rotating sectors of various shapes have been successfully utilized to reduce the difficulties arising from the wide range of the scattering intensity of electrons due to the rapid intensity decrease with scattering angle (Finbak, Hassel \& Ottar, 1941; Hassel \& Viervoll, 1947; Karle \& Karle, 1949). Sectors of the type $s^{n}(n=1,2,3$ and 4) are used to realize an effective exposure time of scattered electrons proportional to $s^{n}$, where $s=4 \pi \sin \theta / \lambda$, $\lambda$ is the wave length and $2 \theta$ is the scattering angle.

To make these sectors, however, necessitates delicate work by hand which requires much skill and time. However, the most essential requirement in using a sector is not that its shape is cut in accordance with the exactly multiple powers of $s$, but that it has a shape which is accurately known. Therefore, if we contrive a simpler shape of sector which can be finished with ordinary cutting machines, we will not only save the laborious work but improve the precision of the sector.

In this paper a new design of sector is described which satisfies the above requirement. This sector, the double circular-dise sector, is composed of two circular dises. Fig. 1 (a) shows schematically the shape of the sector.


Fig. 1.
The two circular dises $A$ and $A^{\prime}$, having the same radius $a$, are in contact at the centre of rotation, $O . O Q$ is the tangent to $A$ and $A^{\prime}$ at $O$. The effective time of exposure for scattered electrons at a point $P$, at distance $r$ from $O$, is proportional to the angle $\theta_{r}$ which is given by

$$
\begin{equation*}
a \sin \theta_{r}=\frac{1}{2} r \tag{1}
\end{equation*}
$$

When $\theta_{r}$ is small, $\sin \theta_{r}$ is nearly proportional to $r$, thus $\theta_{r}$ is approximately a $s^{1}$ sector. At larger value of $\theta_{r}$, however, the deviation from an $s^{1}$ sector is not negligible and equation (1) should be applied.

Fig. 2 is a photograph of the sector. The two discs $A$ and $A^{\prime}$, of radius of about 4 cm ., were cut from a brass plate. The maximum value of $\theta_{r}$ is about $30^{\circ}$. These discs are fixed on a inner race of ball bearing, which is mounted on the holder $H$ with balls between them, and the holder is connected to a shaft $S$ by a spring belt $B$ made of 18-8 stainless steel. The shaft projects from the vacuum through a Wilson seal and is driven by a motor. The sector is rotated at speeds about $200-500$ revolutions per minute. The box under the sector


Fig. . .
carries eight photographic plates $8 \times 12 \mathrm{~cm}$. in size. The upper limit of $s$ on a diffraction photograph is about $30 \AA^{-1}$ for 50 kV . electrons with the camera length 33 cm .

The accuracy of the exposure time, which is limited by the accuracy of cutting the discs and by the vibration due to the rotation, is lower at inner part than at outer part of the sector. In order to attain an accuracy of $1 \%$ at a point 0.5 cm . from the centre, the fluctuation of the radius of the discs, $\Delta r_{1}$, and the radial amplitude of the vibration, $\Delta r_{2}$, should be less than 0.01 mm . and 0.02 mm ., respectively*. The cutting accuracy of 0.01 mm . is easily attained with a lathe. It is more difficult to keep $\Delta r_{2}$ within the limit, especially when the speed of rotation is high; thus, particular care was required in designing the support of the sector. The values of $\Delta r_{2}$ for the present sector was kept under 0.02 mm . for rotation of about 200 revolutions per minute.

In the case of diffraction by crystals and amorphous materials by both reflexion and transmission methods, the background intensities, caused mainly by multiple


Fig. 3. Aperture of sectors as the function of $r$. Broken lines: $s^{1}$ and $s^{2}$ sectors; full line: circular-disc sector of Fig. 1(b). The curves are normalized at $r_{\text {max }}$.

* These requirements are more severe for sectors of higher powers of $s$.
scattering, are superposed on the singly-scattered intensities; and since the variation of the background intensities with scattering angle is in general moderate, the decrease of the total intensity with scattering angles is not so rapid as in the case of diffraction by gas molecules. Thus, the $s^{1}$, or approximately $s^{1}$, sector is useful in the electron-diffraction study of solid and liquid substances.

We can suggest here that another double circular-disc sector, as shown in Fig. l(b) may be useful for samples for which the intensity decrease is more rapid than inverse $s^{1}$. The discs $A$ and $A^{\prime}$ touch at $O$, the centre of rotation. The aperture $\theta_{r}$ of the sector as the function of $r$, is shown in Fig. 3 in comparison with those for the $s^{1}$ and $s^{2}$ sectors. The double circular-dise sector of this type approximates to the $s^{1}$ sector at small values of $r$ and to the $s^{2}$ sector at large values of $r$.

The author expresses her sincere thanks to Prof.
S. Miyake and Dr G. Honjo for their kind discussions, and to the Akashi Manufacturing Company and to Mr T. Masuda of the same Company for making this diffraction camera. Her thanks are also due to Mr I. Yamaguchi*, Canon Camera Company, for his help in designing the sector, and to Mr M. Mitsui, the Head of Chemical Division in the same Company, for his encouragement.

## References

Finbak, C., Hassel, O. \& Ottar, B. (1941). Arch. Math. Naturv. B, 44, No. 13.
Hassel, O. \& Viervoll, H. (1947). Acta Chem. Scand. 1, 149.
Karle, I. L. \& Karle, J. (1949). J. Chem. Phys. 17, 1052.

* Now at University of Electro-Communications.

Acta Cryst. (1957). 10, 709
Crystallographic properties of diammonium phosphate, $\left(\mathbf{N H}_{4}\right)_{2} \mathbf{H P O}_{4}$. By J. P. Smitr, J. R. Lehr and W. E. Brown, Division of Chemical Development, Tennessee Valley Authority, Wilson Dam, Ala., U.S.A.
(Received 19 July 1957)

A rapidly growing interest in diammonium phosphate as a nitrogen-phosphorus fertilizer material prompted a study of its crystallographic properties, which were but partly described in the literature.

Diammonium phosphate forms colorless monoclinic crystals, holohedral class $2 / m$. Crystals from the system $\mathrm{NH}_{3}-\mathrm{H}_{3} \mathrm{PO}_{4}-\mathrm{H}_{2} \mathrm{O}$ usually are tabular to platy on (001), the common forms being $\{001\}$, $\{110\}$ and $\{\overline{1} 01\}$. The presence of nitrate ion causes elongation along $b$ to produce blade-like crystals. Sulfate ion causes thickening along $c$ to produce equant, distorted bipyramids, the forms $\{001\},\{110\}$ and $\{\overline{1} 01\}$ being equally developed. The presence of ferrocyanide ion along with sulfate ion similarly causes thickening along $c$ to give equant crystals but changes the habit to parallelepipedons with equally developed $\{001\}$ and $\{110\}$, sometimes modified by prominent $\{\overline{2} 01\}$. Certain aliphatic surfactants introduce strain, with resultant warping of the tabular crystals on (001).

The crystals are biaxial positive, $2 V=80-85^{\circ}$ (calc. $=85^{\circ}$ ),

$$
N_{\alpha}=1.508, \quad N_{\beta}=1.518, \quad N_{\gamma}=1.530
$$

$B x_{a}=Z=b$, and $X \wedge a=43.5^{\circ}$ in acute $\beta$. Interfacial angle measurements yield $\beta=113 \cdot 5^{\circ}$. No dispersion was observed.

Rotation and Weissenberg photographs (Cu $K \alpha, \lambda=$ $1.54 \AA$ ) show the unit cell to be monoclinic with the dimensions

$$
a=8.03, \quad b=6.68, \quad c=11.02 \AA, \quad \beta=113^{\circ} 38^{\prime}
$$

The only systematic absences apparently are $h 0 l$ with $h$ odd and $0 k 0$ with $k$ odd. The most probable space
group is $C_{2 h}^{5}-P 2_{1} / a$. With $4\left[\left(\mathrm{NH}_{4}\right)_{2} \mathrm{HPO}_{4}\right]$ per unit cell, the calculated specific gravity, $1 \cdot 61$, agrees quite well with the value 1.62 reported by Schiff (1859) and the value 1.60 calculated from refractive indices through the Gladstone-Dale relationship (Larsen \& Berman, 1934).

The comparison in Table 1 shows that the X -ray data
Table 1. Comparison of $X$-ray and optical goniometric results

|  | X-ray | Mitscherlich | Brooke |
| :---: | :---: | :---: | :---: |
| $a: b: c$ | $1 \cdot 2021: 1: 1 \cdot 6497$ | $1 \cdot 1981: 1: 1 \cdot 6546$ | - |
| $(100) \wedge(001)$ | $113^{\circ} 38^{\prime}$ | $113^{\circ} 14^{\prime}$ | - |
| $(110) \wedge(110)$ | $95^{\circ} 23^{\prime}$ | $95^{\circ} 30^{\prime}$ | $95^{\circ} 45^{\prime}$ |
| $(10) \wedge(001)$ | $74^{\circ} 20^{\prime}$ | $74^{\circ} 37^{\prime}$ | $74^{\circ} 10^{\prime}$ |
| $(\overline{1} 01) \wedge(001)$ | $70^{\circ} 21^{\prime}$ | $70^{\circ} 16^{\prime}$ | $70^{\circ} 28^{\prime}$ |

yield axial ratios and interfacial angles in good agreement with the optical goniometric data of Mitscherlich and of Brooke as reported by Groth (1908).

The powder diffraction pattern reported by Hanawalt, Rinn \& Frevel (1938) is suitable as a means of identification, although these authors listed only the stronger lines.

## References

Groth, P. (1908). Chemische Krystallographie, vol. 2. Leipzig: Englemann.
Hanawalt, J. D., Rinn, H. W. \& Frevel, L. K. (1938). Industr. Engng. Chem. Anal. Ed. 10, 474.
Larsen, E. S. \& Berman, H. (1934). Bull. U.S. Geol. Surv. No. 848, 2nd ed.
Schiff, H. (1859). Liebigs Ann. 112, 88.

